

Polymer Bound 3-Hydroxy-2-methylidenpropionic Acids. A Template for Multiple Core Structure Libraries

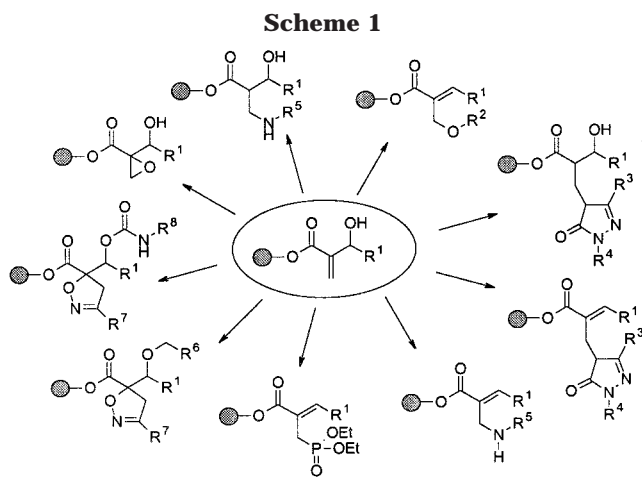
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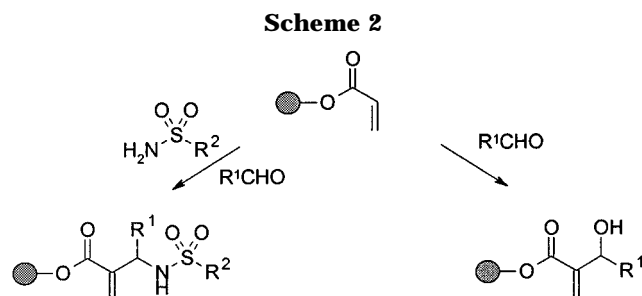
Combinatorial chemistry¹ has become an important tool for the synthesis of small organic molecules which may serve as new lead structures in the drug discovery process. Many of these libraries are based on the solid phase synthesis of a single core structure bearing different substituents. However, for random screening it is important to have structurally different cores containing different commercially available building blocks. Therefore, the synthesis of flexible templates is very promising, since not only the substituents of one core molecule can be varied, but also the type of core derived from this template.

Tempest and Armstrong² have successfully shown the use of squaric acid as a template for multiple core structure libraries (MCSL). Here we present polymer bound 3-hydroxy-2-methylidenpropionic acid as an excellent template for the synthesis of MCSL's. In only a few steps a significant number of different linear and heterocyclic cores can be constructed (Scheme 1).



Polymer bound acrylic ester is reacted in a Baylis–Hillman reaction³ with aldehydes to form 3-hydroxy-2-methylidenpropionic acids⁴ or with aldehydes and sul-

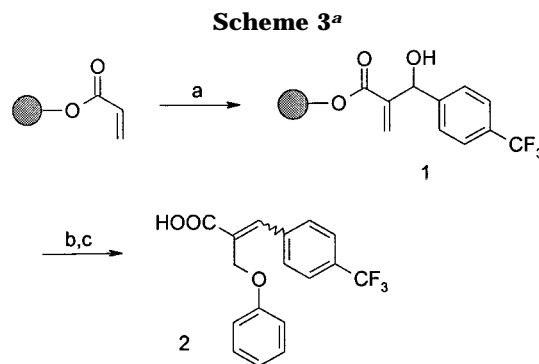
fonamides in a three-component reaction to form 2-methylidene-3-[(arylsulfonyl)amino]propionic acids⁵ (Scheme 2).



Though both structures can be used as templates for the synthesis of MCSL's, in this paper we demonstrate the application of the resin bound 3-hydroxy-2-methylidenpropionic acids. Due to the presence of two functional groups, there is a wide range of possible postmodifications. One possibility is the conversion of the double bond, as in the Michael addition of amines leading to 1,3-amino alcohols,⁴ by epoxidation or a Michael addition of a β -keto ester to form pyrazolones after the conversion into the hydrazones followed by an intramolecular cyclization.

Another possibility is to carry out further reactions involving the hydroxy function, for example in a Mitsunobu reaction, whereby the double bond is rearranged in a manner similar to the synthesis of allylic amines.

The final option is to convert both functional groups in one reaction step. During the synthesis of isoxazolines⁶ via a 1,3-dipolar cycloaddition with nitrile oxides,⁷ the hydroxy function reacts with phenyl isocyanate which is used to dehydrate the nitroalkane to form a carbamate.



^a Key: (a) 16 equiv of 4-trifluoromethylbenzaldehyde, 10 equiv of DABCO, CHCl₃/DMSO 1:1 rt, 2 d, double coupling; (b) 6 equiv of phenol, 3 equiv of DIAD (diisopropyl diazodicarboxylate), 3 equiv of PPh₃, THF, rt, 3 h; (c) 5% TFA/CH₂Cl₂, rt, 30 min.

The starting point for all of these reactions is the Baylis–Hillman reaction itself. For this reaction we chose 4-(trifluoromethyl)benzaldehyde (Scheme 3) and the re-

(1) Früchtel, J. S.; Jung, G. *Angew. Chem., Int. Ed. Engl.* **1996**, *35*, 17–42. Hermkens, P. H. H.; Ottenheijm, H. C. J.; Rees, D. *Tetrahedron* **1997**, *53*, 5643–5678. Brown, R. *Contemp. Org. Synth.* **1997**, 216–237.

(2) Tempest, P. A.; Armstrong, R. W. *J. Am. Chem. Soc.* **1997**, *119*, 7607–7608.

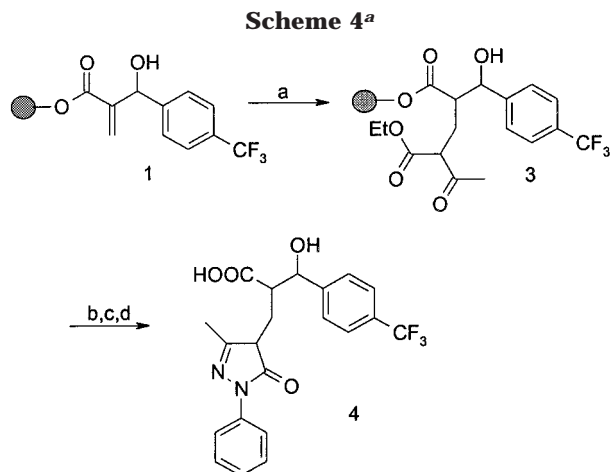
(3) Basavaiah, D.; Rao, P. D.; Hyma, R. S. *Tetrahedron* **1996**, *52*, 8001–8062.

(4) Richter, H.; Jung, G. *Mol. Div.* **1998**, *3*, 191–194. See also: Prien, O. Rölfing, K.; Thiel, A.; Künzer, H. *Synlett* **1997**, 325–326.

(5) Richter, H.; Jung, G. *Tetrahedron Lett.* **1998**, *39*, 2729–2730.

(6) Richter, H.; Höltzel, A.; Jung, G. In *Innovation and Perspectives in Solid-Phase Synthesis and Combinatorial Libraries*; Epton, R., Ed.; Mayflower Scientific Limited: Birmingham, in press.

(7) Mukaiyama, T.; Hoshino, T. *J. Am. Chem. Soc.* **1960**, *82*, 5339–5340.



^a Key: (a) 5 equiv of ethyl acetoacetate, 2 equiv of BEMP, THF, rt, 4 h; (b) 10 equiv of phenylhydrazine hydrochloride, 10 equiv of DIEA, 20 equiv of TMOF, THF, rt, 8 h; (c) 5% TFA/CH₂Cl₂, rt, 30 min.

sulting polymer bound allylic alcohol **1** in order to establish the synthesis of the different cores.

First we carried out the Mitsunobu reaction⁸ with phenol under standard reaction conditions (Scheme 3). This reaction proceeded very well, and we obtained only the S_N'-reaction product **2**.⁹

The incorporation of diverse aromatic and heteroaromatic aldehydes and diverse substituted phenols gave excellent purities (65–98%) as determined by analytical HPLC ($\lambda = 214$ nm).

To show the possibility of Michael additions, we chose the synthesis of pyrazolones¹⁰ (Scheme 4). The Michael addition was carried out with ethyl acetoacetate and BEMP¹¹ as base to form the resin bound β -keto ester **3**. This was then transformed into the hydrazone with phenylhydrazine hydrochloride in the presence of TMOF and DIEA. During cleavage with 5% TFA in CH₂Cl₂ the hydrazone cyclized to form the pyrazolone **4** with a purity of 81% (analytical HPLC) and an isolated yield of 80%.

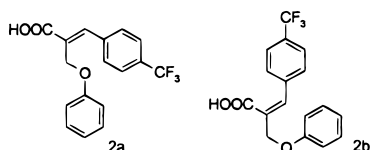
Following this procedure 30 different pyrazolones were synthesized in good purities (44–85%).

To allow for a diverse multistep synthesis we transformed the 3-hydroxy-2-methylidenepropionic acid **1** (Scheme 5) into the polymer bound allylic amine **5**, an unusual β -amino acid. The polymer bound allylic alcohol **1** was first treated with acetyl chloride and DIEA in CH₂Cl₂ to form the ester, which was reacted with cyclopropylamine in an addition elimination step to form the allylic amine **5**.¹² After cleavage from the resin, **5** was obtained in good purity (87%) and yield (76%).

The polymer bound allylic amines **5** can be employed

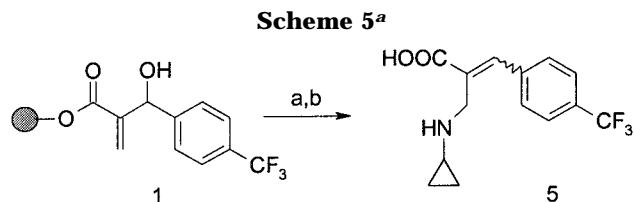
(8) Mitsunobu, O. *Synthesis* **1981**, 1–28.

(9) NMR analysis using NOE difference spectroscopy shows that only **2a** was formed.

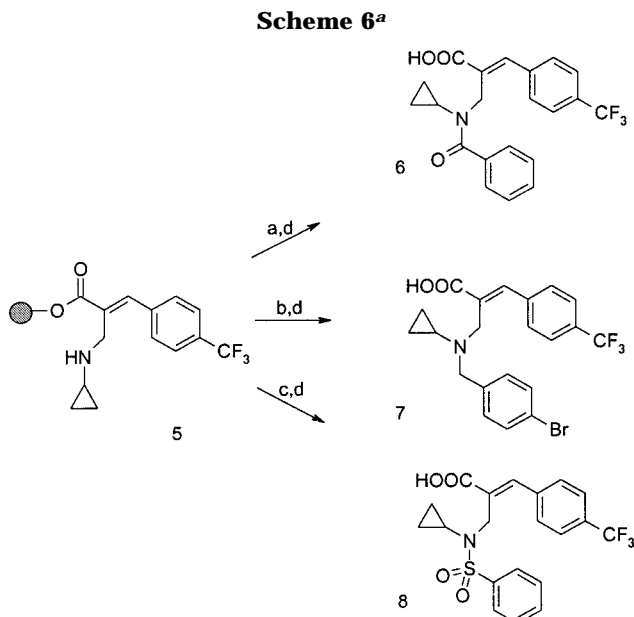


(10) Tietze, L. F.; Steinmetz, A. *Synlett* **1996**, 667–668.

(11) 2-(*tert*-Butylimino)-2-(diethylamino)-1,3-dimethyl-perhydro-1,3,2-diazaphosphorine; Schwesinger, R.; Schlemper, H. *Angew. Chem., Int. Ed. Engl.* **1987**, *26*, 1212–1214.



^a Key: (a) 10 equiv of acetyl chloride, 10 equiv of DIEA, CH₂Cl₂, rt, 16 h; (b) 10 equiv of cyclopropylamine, 2 equiv of BEMP, DMF, rt, 6 h.

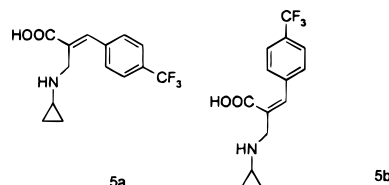


^a Key: (a) 5 equiv of benzoic acid, 5 equiv of PyBroP, 10 equiv of DIEA, CH₂Cl₂/DMF, 1:1, rt, 4 h, double coupling; (b) 10 equiv of 4-bromobenzyl bromide, 4 equiv of BEMP, DMF, rt, 16 h; (c) 10 equiv of benzenesulfonyl chloride, 10 equiv of DIEA, CH₂Cl₂, rt, 5 h; (d) 5% TFA/CH₂Cl₂, 30 min.

in several reactions. Scheme 6 shows acylation, alkylation, and sulfonation as possible modifications. For the acylation we used benzoic acid which was coupled for 3 h by using PyBroP¹³ and DIEA in CH₂Cl₂/DMF 1:1. To drive the reaction to completion, a double coupling was necessary, and after cleavage we obtained the product **6** in high purity (89%) and yield (75%). Alkylation was performed with 4-bromobenzyl bromide and BEMP in DMF for 16 h. In this way we obtained the product **7** in high purity (87%) and yield (80%). The sulfonation was optimized with benzenesulfonyl chloride. The reaction was carried out with DIEA in CH₂Cl₂ for 5 h. After cleavage from the resin product **8** was obtained in high purity (84%) and yield (70%).

To investigate the use of different building blocks we applied 10 different examples of each building block and found good to excellent purities (60–92%) in the analytical HPLC ($\lambda = 214$ nm).

(12) NMR analysis using NOE difference spectroscopy shows that only **5a** was formed.



(13) Bromotris(pyrrolidino)phosphonium hexafluorophosphate.

In summary we have shown the use of 3-hydroxy-2-methylidenepropionic acid as a template for the synthesis of multiple core structure libraries. In this case many reactions are carried out, e.g. the Mitsunobu reaction, the synthesis of pyrazolones, and the synthesis of allylic amines, which can be acylated, alkylated, and sulfonated. These reactions are only a small selection of the possibilities with this molecule and show the advantages of flexible templates.

Experimental Section

Reagents were purchased from Aldrich, Merck, Fluka, Bachem, and Novabiochem. 2-Chlorotrityl chloride resin (1.3 mmol/g, 1% DVB, 200–400 mesh) was purchased from Senn Chemicals. All yields reported are based on the initial loading of the resin.

¹H NMR and ¹³C NMR spectra were recorded at a proton frequency of 250 MHz. NOE difference experiments were carried out at a proton frequency of 600.13 MHz. All samples were dissolved in 500 μ L of DMSO-*d*₆.

Analytical HPLC was performed on a Waters instrument using a Grom NucleoSil C₁₈ column (18 cm, 5 μ m C₁₈, 0.3 mL/min, gradient from 90% aqueous media (0.1% TFA) to 100% CH₃CN (0.1% TFA)) with UV detection at 214 nm. High-resolution spectra were acquired on a Fourier-transform-ion-cyclotron-resonance (FT-ICR) mass spectrometer. The negative ions are generated with an external electrospray source (ESI). The LC-MS studies were performed with a ESI-mass spectrometer coupled with an ABI 140 solvent delivery system. For UV-detection a linear UVIS 214 photometer was used at 214 nm.

3-Hydroxy-2-methylidene-3-[4-(trifluoromethyl)phenyl]propionic Acid (1). 2-Chlorotrityl chloride PS resin (40 mg, capacity: 1.3 mmol/g) was loaded with 4 equiv of acrylic acid (14.3 μ L) in 600 μ L of DMF/DCM (1:1) by using 6 equiv of Et₃N (43.5 μ L) for 4 h at 0 °C. After washing, the resin was suspended in 300 μ L of DMSO/CHCl₃ (1:1) and 16 equiv of 4-(trifluoromethyl)benzaldehyde (111.2 μ L) and 10 equiv of DABCO (58 mg) were added and shaken for 2 d at room temperature. This step was repeated once. The resulting mixture was filtered and washed with DMF, EtOH, CH₂Cl₂, and Et₂O. **1** was cleaved with 600 μ L of DCM/TFA (95:5) for 45 min and concentrated to dryness. After lyophilization from *tert*-BuOH/H₂O (4:1), we obtained a crude product with a purity of 97% in analytical HPLC. **1** was analyzed by MS and purified by preparative HPLC to obtain 10.9 mg (85% yield): ¹H NMR: δ = 5.50 (s, 1), 5.99 (s, 1), 6.21 (s, 1), 7.51 (d, 2, ³J = 8.24 Hz), 7.67 (d, 2, ³J = 8.24 Hz); ¹³C NMR: δ = 70.1, 123.8, 124.3 (¹J_{CF} = 271 Hz), 124.8, 127.6, 127.6, 144.0, 148.0, 166.8; HRMS (ESI): [M – H][–] calcd for C₁₁H₈O₃F₃: 245.043097, found: 245.043626.

2-(Phenoxymethyl)-3-[4-(trifluoromethyl)phenyl]acrylic Acid (2). Resin **1** (40 mg) reacted with 6 equiv of phenol (29.4 mg), 3 equiv of PPh₃ (40.9 mg), and 3 equiv of DIAD (30.1 μ L) in a Mitsunobu reaction to form **2**. The reaction was carried out in THF for 3 h. The resin with the phenol and PPh₃ was cooled to 0 °C. The DIAD was added, and after 10 min at 0 °C the resin was shaken for 3 h at room temperature. The mixture was filtered and washed with DMF, EtOH, CH₂Cl₂, and Et₂O. **2** was cleaved with 600 μ L of CH₂Cl₂/TFA (95:5) for 45 min and concentrated to dryness. After lyophilization from *tert*-BuOH/H₂O (4:1), we obtained a crude product with a purity of 98% in analytical HPLC. This was analyzed by MS and purified by preparative HPLC to obtain 14.9 mg (89% yield): ¹H NMR: δ = 4.74 (s, 2), 6.94 (m, 1), 6.98 (d, 2, *J* = 7.93 Hz), 7.70 (d, 2, *J* = 8.24 Hz), 7.79 (d, 2, *J* = 8.24 Hz), 8.00 (s, 1); ¹³C NMR: δ = 62.4, 114.7, 115.7, 121.0, 124.3 (¹J_{CF} = 271 Hz), 125.5, 129.5, 130.1, 130.2, 138.4, 142.1, 158.1, 167.7; HRMS (ESI): [M – H][–] calcd for C₁₇H₁₂O₃F₃: 321.074396, found: 321.085600.

2-Acetyl-4-[[hydroxy-4-(trifluoromethyl)phenyl]methyl]pentanedioic Acid 5-Ethyl Ester (3). Resin **1** (40 mg) was suspended in 700 μ L of THF. Then 5 equiv of ethyl acetoacetate (33 μ L) and 2 equiv of BEMP (30.1 μ L) were added and shaken for 5 h at room temperature. After the resin was washed, **3** was cleaved with 600 μ L of DCM/TFA (95:5) for 45 min and concentrated to dryness. After lyophilization from *tert*-BuOH/H₂O (4:1), the crude product had a purity of 80% (summary of

all diastereomers) in analytical HPLC. The identity of **3** was confirmed by MS and HPLC-MS, and **3** was purified by preparative HPLC to obtain 14.2 mg (75% yield): ¹H NMR of one diastereomer: δ = 1.21 (t, 3, *J* = 7.0 Hz), 2.24 (s, 1), 2.99 (m, 1), 4.1 (m, 2), 5.17 (d, 1, *J* = 8.24 Hz), 7.60 (d, 2, *J* = 8.24 Hz), 7.77 (d, 2, *J* = 8.24 Hz); ¹³C NMR of one diastereomer: δ = 14.2, 19.5, 24.5, 43.3, 59.5, 77.5, 100.4, 124.3 (¹J_{CF} = 271 Hz), 125.3, 127.9, 143.1, 163.1, 166.7, 172.93; HRMS (ESI): [M – H][–] calcd for C₁₇H₁₈O₆F₃: 375.106086, found: 375.127149.

3-Hydroxy-2-[(3-methyl-5-oxo-1-phenyl-4,5-dihydro-1H-pyrazol-4-yl)methyl]-3-[4-(trifluoromethyl)phenyl]propionic Acid (4). To resin **3** (40 mg) were added 10 equiv of phenylhydrazine hydrochloride (75.2 mg), 10 equiv of DIEA (89 μ L), and 20 equiv of TMOF (113.9 μ L) and suspended in 700 μ L of THF. After the mixture was shaken for 16 h at room temperature, it was filtered and washed with DMF, EtOH, CH₂Cl₂, and Et₂O. During cleavage from the resin with 600 μ L of CH₂Cl₂/TFA (95:5), **4** was formed by acid-catalyzed intramolecular cyclization within 30 min. After lyophilization from *tert*-BuOH/H₂O, we obtained a crude product with a purity of 81% (summary of all diastereomers) in analytical HPLC (λ = 214 nm). **4** was analyzed by MS and purified by preparative HPLC to obtain 17.5 mg (80% yield): ¹H NMR of one diastereomer: δ = 1.97/2.04 (m, 2), 2.04 (s, 3), 2.92 (m, 1), 4.73 (d, 1, *J* = 8.4 Hz), 7.16 (t, 1, *J* = 7.6 Hz), 7.40 (t, 2, *J* = 7.6 Hz), 7.60 (d, 2, *J* = 8.24 Hz), 7.67 (d, 2), 7.71 (d, 2, *J* = 8.24 Hz); ¹³C NMR of one diastereomer: δ = 11.2, 21.9, 52.1, 73.6, 102.2, 118.6, 124.3 (¹J_{CF} = 271 Hz), 124.3, 124.8, 127.6, 128.8, 137.7, 147.8, 148.3, 174.6; HRMS (ESI): [M – H][–] calcd for C₂₁H₁₈O₄N₂F₃: 419.122404, found: 419.150434.

2-[(Cyclopropylamino)methyl]-3-[4-(trifluoromethyl)phenyl]acrylic Acid (5). To resin **1** (40 mg) were added 6 equiv of acetyl chloride (22.2 μ L) and 10 equiv of DIEA (89 μ L) in 700 μ L of CH₂Cl₂. The reaction mixture was shaken for 3 h at room temperature. After the resin was washed, 10 equiv of cyclopropylamine (36.5 μ L) and 2 equiv of BEMP (30.1 μ L) in 700 μ L of DMF were added. The reaction mixture was shaken for 4 h at room temperature and then filtered and washed with DMF, EtOH, CH₂Cl₂, and Et₂O. **5** was cleaved from the resin with 600 μ L of CH₂Cl₂/TFA (95:5) in 30 min. After lyophilization from *tert*-BuOH/H₂O, we obtained a crude product with a purity of 87% in analytical HPLC (λ = 214 nm). This was purified by preparative HPLC to obtain 12.2 mg (76% yield): ¹H NMR: δ = 0.67/0.75 (m, 4), 2.68 (m, 1), 4.00 (s, 2), 7.72 (d, 2, *J* = 8.24 Hz), 7.84 (d, 2, *J* = 8.24 Hz), 8.03; ¹³C NMR: δ = 3.4, 30.4, 44.1, 124.28 (¹J_{CF} = 271 Hz), 125.5, 126.8, 130.0, 138.0, 143.3, 167.4; HRMS (ESI): [M – H][–] calcd for C₁₄H₁₃O₂NF₃: 284.090379, found: 284.089488.

2-[(Benzoylcyclopropyl)amino]methyl-3-[4-(trifluoromethyl)phenyl]acrylic Acid (6). To resin **5** (40 mg) were added 5 equiv of benzoic acid (31.7 mg), 5 equiv of PyBroP (121.2 mg) and 10 equiv of DIEA (89 μ L) and suspended in 700 μ L of CH₂Cl₂/DMF (1:1). After the mixture was shaken for 4 h at room temperature and repetition of the procedure, it was filtered and washed with DMF, EtOH, CH₂Cl₂, and Et₂O. **6** was cleaved from the resin with 600 μ L of CH₂Cl₂/TFA (95:5) in 30 min. After lyophilization from *tert*-BuOH/H₂O, we obtained a crude product with a purity of 89% in analytical HPLC (λ = 214 nm). This was analyzed by MS and purified by preparative HPLC to obtain 15.2 mg (75% yield): ¹H NMR: δ = 0.18/0.31 (m, 4), 2.46 (m, 1), 4.53 (s, 2), 7.13 (d, 2, *J* = 7.02 Hz), 7.33 (m, 3), 7.70 (d, 2, 8.24 Hz), 7.76 (d, 2, *J* = 8.24 Hz), 7.83 (s, 1); ¹³C NMR: δ = 9.4, 30.8, 43.2, 124.28 (¹J_{CF} = 271 Hz), 125.1, 127.0, 127.7, 129.4, 129.8, 131.3, 137.3, 139.0, 139.2, 168.1, 171.2; HRMS (ESI): [M – H][–] calcd for C₂₁H₁₇O₃NF₃: 388.116592, found: 388.113470.

2-[(4-Bromobenzyl)cyclopropylamino]methyl-3-[4-(trifluoromethyl)phenyl]acrylic Acid (7). To resin **5** (40 mg) were added 10 equiv of 4-bromobenzyl bromide (130 mg) and 4 equiv of BEMP (60.2 μ L) and suspended in 700 μ L of DMF. After the mixture was shaken for 16 h at room temperature, it was filtered and washed with DMF, EtOH, CH₂Cl₂, and Et₂O. **7** was cleaved from the resin with 600 μ L of CH₂Cl₂/TFA (95:5) in 30 min. After lyophilization from *tert*-BuOH/H₂O, we obtained a crude product with a purity of 87% in analytical HPLC (λ = 214 nm). **7** was analyzed by MS and purified by preparative HPLC to obtain 18.8 mg (80% yield): ¹H NMR: δ = 0.51 (m, 4), 2.06 (m, 1), 3.73 (s, 1), 3.94 (s, 2), 7.27 (d, 2, *J* = 8.24 Hz), 7.52 (d, 2, *J* = 8.24

Hz), 7.64 (m, 4), 7.82 (s, 1); ^{13}C NMR: $\delta = 6.4, 37.0, 50.1, 59.2, 121.4, 124.3$ ($^1J_{\text{CF}} = 271$ Hz), 125.2, 128.6, 129.1, 130.2, 131.1, 132.6, 138.6, 141.0, 168.6; HRMS (ESI): $[\text{M} - \text{H}]^-$ calcd for $\text{C}_{21}\text{H}_{18}\text{O}_2\text{NF}_3\text{Br}$: 452.047838/454.045792, found 452.044660/454.043447.

2-[(Benzenesulfonylcyclopropylamino)methyl]-3-[4-(trifluoromethyl)phenyl]acrylic Acid (8**).** To resin **5** (40 mg) were added 10 equiv of benzenesulfonyl chloride (66.7 mg) and 10 equiv of DIEA (89 μL) and suspended in 700 μL of CH_2Cl_2 . After the mixture was shaken for 4 h at room temperature, it was filtered and washed with DMF, EtOH, CH_2Cl_2 , and Et_2O . **8** was cleaved from the resin with 600 μL of $\text{CH}_2\text{Cl}_2/\text{TFA}$ (95:5) in 30 min. After lyophilization from *tert*-BuOH/ H_2O we obtained a crude product with a purity of 84% in analytical HPLC ($\lambda = 214$ nm). This was purified by preparative HPLC to obtain 15.5 mg (70% yield): ^1H NMR: $\delta = 0.61/0.78$ (m, 4), 1.62 (m, 1), 4.06 (s, 2), 7.67 (m, 9), 7.8 (s, 1); ^{13}C NMR: $\delta = 7.4, 31.2, 46.8, 124.3$

($^1J_{\text{CF}} = 271$ Hz), 125.2, 127.8, 129.2, 130.3, 130.7, 133.3, 135.1, 138.7, 140.2, 166.1; HRMS (ESI): $[\text{M} - \text{H}]^-$ calcd for $\text{C}_{20}\text{H}_{17}\text{NO}_4\text{SF}_3$: 424.083576, found: 424.080076.

Acknowledgment. This work was supported by the BMBF grant 03D0037A7.

Supporting Information Available: MS, ^1H NMR, and ^{13}C NMR spectra of all products; NOE difference spectra of compounds **2a** and **5a**; HPLC chromatograms of all products and HPLC-MS spectra of compound **3** and **4**. Tables with HPLC purities of the building blocks used in the presented syntheses (39 pages). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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